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(2.1), (2.2) Evaluation Ask students to discuss the probability of life on a planet that is not abundant in the elements that form covalent bonds. (2.2) Ask students to explain which characteristics of life mentioned in Chapter 1 are determined by the properties of elements making up organisms. (2.1), (2.2), (2.3), (2.4)

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(2.2) In a small class setting, bring in samples of polar and nonpolar substances and mix them together. In a large class, use an overhead or videocam with LCD setup to project it to the entire class; this may take a little ingenuity when working on a horizontal surface.

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Brooker properties of glue. (2.3, 2.4) Ask students to explain why the digestive systems of animals must be adapted to break down covalent bonds yet there is no particular system for breaking down ionic bonds. (2.2) Ask students to explain why pH is a factor used Page 7/21

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(2.2) x Ask students to explain why pH is a factor used in food preservation.(2.4) Analysis x Ask students to explain what types of organisms would be most affected

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At the University of Minnesota, Dr. Brooker teaches undergraduate courses in biology, genetics, and cell biology. In addition to many other publications, he has written two undergraduate genetics texts published by McGraw-Hill: Genetics: Analysis & Principles, 4th edition, copyright 2012, and Concepts of Genetics, copyright 2012.

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