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Solution. Physical: boiling and melting are physical changes. When water boils no bonds are broken or formed. The change could be written: $\text{H}_2\text{O} (l) \rightarrow \text{H}_2\text{O} (g)$ Chemical: The dark grey nail changes color to form an orange flaky substance (the rust); this must be a chemical change.

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Supplemental Problems Chemistry: Matter and Change • Chapter 2 1 Data Analysis
Data Analysis 1. A sample of aluminum is placed in a 25-mL graduated cylinder containing 10.0 mL of water. The level of water rises to 18.0 mL. Aluminum has a density of 2.7 g/mL. Calculate the mass of the sample. 2. Saturn is about 1 429 000 km from the Sun.

3.6: Changes in Matter: Physical and Chemical Changes ...

2 Chemistry: Matter and Change Solving Problems: A Chemistry Handbook
SOLVING PROBLEMS: CHAPTER 1 A CHEMISTRY HANDBOOK Matter is made up of particles, called atoms, that are so small they cannot be seen with an ordinary light

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microscope. The structure, composition, and behavior of all matter can be explained by atoms and the changes they undergo.

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1.4: Chemistry - A Study of Matter and Its Changes ...

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Chemistry: Matter & Change, Solutions Manual

Chemical properties of matter describes its "potential" to undergo some chemical change or reaction by virtue of its composition. What elements, electrons, and bonding are present to give the potential for chemical change. It is quite difficult to

define a chemical property without using the word "change".

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