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### Skill Practice 10 - S.W.H.S CHEMISTRY

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An element exists as 4 different isotopes. 4.35% have a mass of 49.9461, 83.79% have a mass of 51.9405, 9.50% have a mass of 52.9407 amu, and 2.36% have a mass of 53.9389 amu. a. What is the average atomic mass for this element? b. What is the identity of this element?

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This collection of chemistry test questions is grouped according to subject. Each exam has answers supplied at the end. They provide a useful study tool for students. For instructors, they are a good resource for homework, quiz, or test questions, or practice for the AP Chemistry test.

### Answers To Chemistry Skill Practice 16

Skill Practice 3 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ 1. Explain why compounds are always homogeneous, but mixtures can be either homogeneous or heterogeneous. With a compound (such as salt or sugar), one microscopic portion has the same composition as another microscopic portion.

### Chemistry Textbooks :: Homework Help and Answers :: Slader

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PLEASE NOTE: If you have a question about these answers, it is your responsibility to come to office hours or ask during class work time. UNIT 12 - HW Practice Keys - ChemActivity 51: Cell Voltage - ChemActivity 50: Electrochemical Cell UNIT 11 - HW Practice Keys - ChemQuest 55: Free Energy - ChemQuest 54: 2nd Law of Thermodynamics

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### **Chem Quest - Mr. Smith**

Skill Practice 10 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ 1. Define the terms "ground state" and "excited state". Ground state:  
the normal energy level that an electron occupies. Excited state: when an electron has absorbed energy to occupy a higher  
energy level. 2. What is the wavelength of ...

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